

# SMART SAYLLABUS

MATHS, PHYSICS, CHEMISTRY, BIOLOGY

Class 10th (KPK)

NAME: \_\_\_\_\_

F.NAME: \_\_\_\_\_

CLASS: \_\_\_\_\_ SECTION: \_\_\_\_\_

ROLL #: \_\_\_\_\_ SUBJECT: \_\_\_\_\_

ADDRESS: \_\_\_\_\_

\_\_\_\_\_

SCHOOL: \_\_\_\_\_



<https://web.facebook.com/TehkalsDotCom/>



<https://tehkals.com/>



### Smart Syllabus 10<sup>th</sup> Class Mathematics (KPK Science) 2020-21

Exercise 1.1 (Q.1, 2, 3)	Exercise 4.1 (Q.1, 3, 4, 6, 7, 9, 10)	Theorem 8.1
Exercise 1.2 Q.1 (i), (iii), (iv), (vi), (vii), (viii), (xi), (xiii)	Exercise 4.2 (Q.1, 3, 4, 7, 9, 10)	Review Exercise 8 (Q.1)
Exercise 1.3 Q.1 (i), (iv), (v), (vii), (ix)	Review Exercise 4 (Q.1)	Theorems 9.1, 9.2, 9.3, 9.4 and 9.5
Q.2	Exercise 5.1 (Q.1, 2, 4)	Review Exercise 9 (Q.1)
Review Exercise 1 (Q.1)	Exercise 5.2 (Q.1, 2, 3, 4)	Theorems 10.1, 10.2 and 10.3
Exercise 2.1 (Q.1, 2, 3)	Exercise 5.3 (Q.2, 3, 5)	Review Exercise 10 (Q.1)
Exercise 2.2 (Q.1, 2, 3)	Exercise 5.4 (Q.1, 2, 4)	Theorems 11.1 Case (a), 11.2 Case (a), 11.3 Case (b) and 11.4 Case (b)
Exercise 2.3 (Q.1, 2, 3, 5, 6, 7)	Exercise 5.5 (Q.1, 4, 5, 6)	Review Exercise 11 (Q.1)
Exercise 2.4 (Q.2, 4, 5, 6)	Review Exercise 5 (Q.1)	Theorems 12.3 (a), 12.3 (b) and 12.3 (c)
Exercise 2.5 (Q.1, 2, 4, 6)	Exercise 6.1 (Q.1, 3, 4)	Review Exercise 12 (Q.1)
Exercise 2.6 Q.1 (i), (iv), (vi), (vii)	Exercise 6.2 (Q.1, 2, 4, 6)	Exercise 13.1 (Q.2, 4, 5, 6, 9)
Exercise 2.7 (Q.1, 2, 5, 6, 8, 9)	Exercise 6.3 (Q.2, 3, 4, 6, 7, 8, 9)	Exercise 13.2 (Q.1, 2, 3, 4, 5, 6, 7)
Review Exercise 2 (Q.1)	Exercise 6.4 (Q.1, 3, 4, 5, 6)	Exercise 13.3 (Q.1, 3, 4, 5, 6, 8)
Exercise 3.1 (Q.1, 2, 3, 4, 5, 6)	Review Exercise 6 (Q.1)	Review Exercise 13 (Q.1)
Exercise 3.2 (Q.1, 2, 4, 7)	Exercise 7.1 (Q.1, 2, 3, 4)	
Exercise 3.3 (Q.1, 3, 5, 7)	Exercise 7.2 (Q.1, 2, 3, 5, 6, 7, 8)	
Exercise 3.4 (Q.1, 2, 3, 5, 6, 7)	Exercise 7.3 (Q.1, 2)	
Exercise 3.5 (Q.1, 2, 3, 4, 5)	Exercise 7.4 (Q.1, 2, 3, 4, 5, 6)	
Review Exercise 3 (Q.1)	Exercise 7.5 (Q.1, 2, 5, 6, 11, 12, 13, 14, 15, 16, 17)	
	Exercise 7.6 (Q.1, 3, 4, 7)	
	Review Exercise 7 (Q.1)	



--	--	--

**Table of Test Specification (Sample)**

S. No.	Unit	Percentage	Section-A		Section-B		Section-C	
			No. of MCQs	Marks	No. of Questions	Marks	No. of Questions	Mark
1	Quadratic Equations	7%	1	1	1	4	-	-
2	Theory of Quadratic Equations	13%	1	1	2	8	-	-
3	Variations	6%	1	1	1	4	-	-
4	Partial Fractions	5%	1	1	1	4	-	-
5	Sets and Functions	12%	1	1	2	8	-	-
6	Basic Statistics	10%	2	2	1	4	-	-
7	Introduction to Trigonometry	11%	2	2	1	4	-	-
8	Projection of a Side of a Triangle	6%	1	1	1	4	-	-
9	Chords of a Circle	5%	1	1	-	-	1	8
10	Tangent to a Circle	6%	1	1	-	-	1	8
11	Chords and Arcs	5%	1	1	1	4	-	-
12	Angle in a Segment of a Circle	6%	1	1	1	4	1	8
13	Practical Geometry - Circle	8%	1	1	-	-	1	8
	<b>TOTL</b>	<b>100%</b>	<b>15</b>	<b>15</b>	<b>12</b>	<b>48</b>	<b>04</b>	<b>32</b>

**Percentage (Cognitive Levels)**

Cognitive Level	Percentage
Knowledge	10%
Understanding	20%
Application	70%

**Percentage (Difficulty Levels)**

Difficulty Level	Weightage
Easy (E)	15%
Moderate (M)	70%
Difficult (D)	15%







**Assessment weightage of Physics Grade –X According to (Curriculum, 2006)**

**SAMPLE: TOS (TABLE OF SPECIFICATION)**

**Table1. Cognitive abilities assessment weightage in Physics Grade-X Theory Paper**

Questions	Knowledge (K)			Understanding (U)			Application (A)		
	Questions	Marks	%	Questions	Marks	%	Questions	Marks	%
MCQs	2	2	17	9	9	75	1	1	8
Short Questions	1	4	10	9	36	80	1	4	10
Long Questions	1	3	11	3+3	9+12	74	1	4	15

**Table 2. Overall assessment Percentage of (Cognitive Levels)**

Cognitive Level	Percentage
Knowledge(K)	20
Understanding(U)	45
Application(A)	35
Total	100

**Table 3. Chapter wise assessment weightage in Physics Grade-X Theory Paper**

Unit No	Unit	Assessment weightage	MCQs Section-A Marks: 12		Short Questions Section-B Option: 8 questions out of 11 Marks:32		Long Questions Section-C Option: 3 questions out of 4 Marks:21	
			Question	Marks	Questions	Marks	Questions	Marks
10	Simple Harmonic Motion and Waves	10	1	1	1	4	1	3
11	Sound	10	1	1	1	4	1	4
12	Geometrical Optics	10	2	2	1	4	1	4
13	Electrostatics	12	1	1	1	4	1	3
14	Current Electricity	14	2	2	2	8	1	4
15	Electromagnetism	14	1	1	2	8	1	3
16	Introductory Electronics	10	2	2	1	4	1	3
17	Information and Communication Technology	10	1	1	1	4	-	-
18	Radioactivity	10	1	1	1	4	1	4
<b>Total</b>	-	<b>100</b>	<b>12</b>	<b>12</b>	<b>11</b>	<b>44</b>	<b>8</b>	<b>28</b>

**Note: This Table of specification is a sample TOS for Physics Grade-X paper.**



**CHEMISTRY GRADE-X**

<p><b>Unit # 9</b> <b>CHEMICAL EQUILIBRIUM</b></p>	<p><b>Unit # 11</b> <b>ORGANIC CHEMISTRY</b></p>
<ul style="list-style-type: none"> <li><input type="checkbox"/> Introduction</li> <li><input type="checkbox"/> Reversible and dynamic equilibrium</li> <li>Graphical representation of dynamic equilibrium and its example</li> <li><input type="checkbox"/> Forward reaction and reverse reaction</li> <li><input type="checkbox"/> Law of mass action</li> <li><input type="checkbox"/> Derivation of the expression for equilibrium constant of a general reaction</li> <li><input type="checkbox"/> Equilibrium constant and its units</li> <li><input type="checkbox"/> Finding values of equilibrium constant</li> </ul>	<ul style="list-style-type: none"> <li><input type="checkbox"/> Molecular formula</li> <li><input type="checkbox"/> Structural formula</li> <li><input type="checkbox"/> Condensed formula</li> <li><input type="checkbox"/> Dot and cross formula</li> <li><input type="checkbox"/> Organic compounds</li> <li><input type="checkbox"/> Classification of organic compounds</li> <li><input type="checkbox"/> Diversity and magnitude of organic compounds</li> <li><input type="checkbox"/> Alkane</li> <li><input type="checkbox"/> Naming of alkane</li> <li><input type="checkbox"/> Homologous series</li> <li><input type="checkbox"/> Cyclo alkanes</li> <li><input type="checkbox"/> Alkyl radicals</li> <li><input type="checkbox"/> Functional group containing C,H and oxygen</li> <li><input type="checkbox"/> Functional group containing C,H and N</li> <li><input type="checkbox"/> Functional group containing C,H &amp; Halogen</li> <li><input type="checkbox"/> Double and triple bond</li> </ul>
<p style="text-align: center;"><b>Exercise</b></p> <ul style="list-style-type: none"> <li><input type="checkbox"/> MCQs: 1,2,3,4,6,8,9,10</li> <li><input type="checkbox"/> Short question: Q1,2,3,5,6,8,9</li> <li><input type="checkbox"/> Long questions: Q1(i),(ii),(iii) Q2(a)&amp;(b) Q3 part (a) only Q4 part(a) only Q5 part(a) only</li> </ul>	<p style="text-align: center;"><b>Exercise</b></p> <ul style="list-style-type: none"> <li><input type="checkbox"/> MCQs: 1,2,3,4,5,6,7,10</li> <li><input type="checkbox"/> Short questions: 1,3,4,6,8,9,10</li> <li><input type="checkbox"/> Long questions: Q1 part (b) only Q3 both (a)&amp;(b) Q4 (a),(b)&amp;(c) Q5 (a),(b)&amp;(c)</li> </ul>
<p style="text-align: center;"><b>Unit # 10</b> <b>ACIDS, BASES AND SALTS</b></p> <ul style="list-style-type: none"> <li><input type="checkbox"/> Concept of Acids and Bases</li> <li><input type="checkbox"/> Arrhenius concept of Acids and Bases</li> <li><input type="checkbox"/> Bronsted and lowery concept</li> <li><input type="checkbox"/> Lewis concept of Acids and Bases</li> <li><input type="checkbox"/> pH and pOH</li> <li><input type="checkbox"/> Example No. 10.2 and 10.3</li> <li><input type="checkbox"/> Neutralization reaction</li> <li><input type="checkbox"/> Salt</li> <li><input type="checkbox"/> Table 10.4</li> <li><input type="checkbox"/> Types of salts</li> </ul>	<p style="text-align: center;"><b>Unit # 12</b> <b>HYDROCARBONS</b></p> <ul style="list-style-type: none"> <li><input type="checkbox"/> Introduction</li> <li><input type="checkbox"/> Alkanes</li> <li><input type="checkbox"/> Table 12.1</li> <li><input type="checkbox"/> Nomenclature</li> <li><input type="checkbox"/> Preparation of alkanes</li> <li><input type="checkbox"/> Hydrogenation of alkenes and alkynes</li> <li><input type="checkbox"/> Reduction of alkyl halides</li> <li><input type="checkbox"/> Physical Properties</li> <li><input type="checkbox"/> Important reactions <ul style="list-style-type: none"> <li>a. Halogenation</li> <li>b. Combustion</li> </ul> </li> <li><input type="checkbox"/> Alkenes</li> <li><input type="checkbox"/> Table 12.2</li> <li><input type="checkbox"/> Rules for Naming of alkenes</li> <li><input type="checkbox"/> Preparation <ul style="list-style-type: none"> <li>a. Dehydration of alcohol</li> <li>b. Dehydrohalogenation of alkyl Halides</li> </ul> </li> <li><input type="checkbox"/> Physical Properties</li> <li><input type="checkbox"/> Important Reactions <ul style="list-style-type: none"> <li>a. Addition of Halogens</li> <li>b. Addition of Hydrogen Halides</li> <li>c. Addition of Hydrogen</li> </ul> </li> <li><input type="checkbox"/> Oxidation with KMnO<sub>4</sub></li> </ul>
<p style="text-align: center;"><b>Exercise</b></p> <ul style="list-style-type: none"> <li><input type="checkbox"/> MCQs: 1,2,3,4,5,6,7,8,9</li> <li><input type="checkbox"/> Short questions: 1,2,4,5,6,8,9,10</li> <li><input type="checkbox"/> Long questions: Q1 (a)&amp;(b) Q3 (a)&amp;(b) Q4 (a),(b)&amp;(c) Q5 (a)&amp;(b)</li> </ul>	



<b>CHEMISTRY GRADE-X</b>	
<p style="text-align: center;"><b>Exercise Unit # 12</b></p> <input type="checkbox"/> MCQs 1 to 10(All) <input type="checkbox"/> Short questions 1,2,,4,5,6,7,8,10 <input type="checkbox"/> Long questions <input type="checkbox"/> Q2 Part (i) only <input type="checkbox"/> Q3 (i) & (ii) <input type="checkbox"/> Q4 (a),(b), (c) & (d) <input type="checkbox"/> Q5 a,b& c only	<p style="text-align: center;"><b>Unit # 14</b></p> <input type="checkbox"/> Ozone depletion and its effects <input type="checkbox"/> Ozone <input type="checkbox"/> Occurrence of Ozone <input type="checkbox"/> Ozone formation <input type="checkbox"/> Ozone layer depletion <input type="checkbox"/> Causes of Ozone Layer depletion <input type="checkbox"/> Ozone Layer depletion <input type="checkbox"/> Due to NOx <input type="checkbox"/> Due to nascent Oxygen <input type="checkbox"/> Due to SOx <input type="checkbox"/> Ways to bring down Ozone Layer depletion <input type="checkbox"/> Effects of global warming
<p style="text-align: center;"><b>Unit # 13</b> <b>BIOCHEMISTRY</b></p> <input type="checkbox"/> Carbohydrates <input type="checkbox"/> Classification <input type="checkbox"/> Sources of Carbohydrates <input type="checkbox"/> Uses of Carbohydrates <input type="checkbox"/> Proteins <input type="checkbox"/> Amino acid as building block of proteins <input type="checkbox"/> Sources and uses of proteins <input type="checkbox"/> Lipids <input type="checkbox"/> Classification <input checked="" type="checkbox"/> Fatty Acids <input type="checkbox"/> Difference between fats and oil <input type="checkbox"/> Sources and uses of lipids	<p style="text-align: center;"><b>Exercise</b></p> <input type="checkbox"/> MCQs 1,3,4,5,6,7,8,9,10 <input type="checkbox"/> Short Question 1,3,4,5,6,7,8,9,10 <input type="checkbox"/> Long Question Q1 (a) & (b) Q3 (a) , (b) & (c) Q4 (a), (b), (c) & (d) Q5 (a), (b) ,(c) & (d)
<p style="text-align: center;"><b>Exercise</b></p> <input type="checkbox"/> MCQs 1,2,3, 6, 7,8,9 <input type="checkbox"/> Short question 1,2,3,4,7,8,10 <input type="checkbox"/> Long question Q1, Q2, Q3	<p style="text-align: center;"><b>Unit # 15</b> <b>ENVIRONMENTAL CHEMISTRY- II</b> <b>WATER</b></p> <input type="checkbox"/> Physical properties of water <input type="checkbox"/> Chemical properties of water <input type="checkbox"/> Thermal Stability of Water <input type="checkbox"/> Reaction with metals <input type="checkbox"/> Non-metals <input type="checkbox"/> Reactions with metallic oxides <input type="checkbox"/> Reactions with non-metallic oxides <input type="checkbox"/> Hydrolysis <input type="checkbox"/> Water as a universal Solvent <input type="checkbox"/> Polarity of Water molecule <input type="checkbox"/> Hydrogen Bonding <input type="checkbox"/> Dielectric constant <input type="checkbox"/> Soft and Hard Water <input type="checkbox"/> Causes of hardness <input type="checkbox"/> Types of Hardness of water a. Temporary Hard Water b. Permanent Hard water <input type="checkbox"/> Methods of removing of Hardness <input type="checkbox"/> Disadvantages of Hard Water
<p style="text-align: center;"><b>Unit # 14</b> <b>ENVIRONMENTAL CHEMISTRY-I;</b> <b>ATMOSPHERE</b></p> <input type="checkbox"/> Atmosphere <input type="checkbox"/> Composition of Atmosphere <input type="checkbox"/> Layers of Atmosphere <input type="checkbox"/> Troposphere <input type="checkbox"/> Stratosphere <input type="checkbox"/> Table 14.2 <input type="checkbox"/> Importance of Atmosphere <input type="checkbox"/> Acid Rain <input type="checkbox"/> Formation of Acid Rain <input type="checkbox"/> Effects of acid Rain on fishes and wild life <input type="checkbox"/> Effects of acid Rain on Plants and Trees <input type="checkbox"/> Effects of Acid Rain on Material <input type="checkbox"/> Effects of Acid Rain on Human <input type="checkbox"/> Effects of Acid Rain on Soil	<p style="text-align: center;"><b>Exercise</b></p> <input type="checkbox"/> MCQs 1,2,3,4,5,7,10 <input type="checkbox"/> Short Questions    1,6,7,8 <input type="checkbox"/> Long Questions Q1 part (b) and (c) only   Q2 Q3 (a,b&c)







**TABLE OF SPECIFICATION FOR CHEMISTRY PAPER GRADE-X**

S#	Unit name	Assessment weightage		MCQs (Total-12)		Short questions (Section-B) (08/11 questions) Marks-32		Long Questions (Section-C) Questions 3/4 Marks-21	
		Curriculum - 2006	Adjust:	Questions	Marks	Questions	Marks	Questions	Marks
1	Chemical equilibrium	5	10	1	1	1	4	1/2 part of long question	3.5
2	Acids ,Bases and Salts	7	14	2	1	2	8	1/2 part of long question	3.5
3	Organic Chemistry	5	10	1	1	1	4	1/2 part of long question	3.5
4	Hydrocarbons	5	10	1	2	1	8	.....	.....
5	Biochemistry	6	12	1	1	1	4	1/2 part of Long question	3.5
6	Environmental Chemistry I; The Atmosphere	7	14	2	2	2	8	1/2 part of long question	3.5
7	Environmental Chemistry II; Water	8	16	2	2	1	4	1/2 part of Long question	3.5
8	Chemical Industries	9	18	2	2	2	8	2/2 part of long question	7
<b>Total</b>		<b>52</b>	<b>104</b>	<b>12</b>	<b>12</b>	<b>11</b>	<b>44</b>	<b>04</b>	<b>28</b>

**Assessment weightage of Chemistry Grade –X According to (Curriculum, 2006) only for Session (2020-2021)**

**TOS (TABLE OF SPECIFICATION)**

**Cognitive Abilities Assessment Weightage in Chemistry Grade-X Theory Paper For Session (2020-2021)**

Questions	Knowledge (K)			Understanding (U)			Application (A)		
	Questions	Marks	%	Questions	Marks	%	Questions	Marks	%
<b>MCQs</b>	<b>2</b>	<b>2</b>	<b>17</b>	<b>9</b>	<b>9</b>	<b>75</b>	<b>1</b>	<b>1</b>	<b>8</b>
<b>Short Questions</b>	<b>1</b>	<b>4</b>	<b>10</b>	<b>9</b>	<b>36</b>	<b>80</b>	<b>1</b>	<b>4</b>	<b>10</b>
<b>Long Questions</b>	<b>1</b>	<b>3</b>	<b>11</b>	<b>3+3</b>	<b>9+12</b>	<b>74</b>	<b>1</b>	<b>4</b>	<b>15</b>

**Overall assessment Percentage of (Cognitive Levels)**

Cognitive Level	Percentage
Knowledge(K)	10
Understanding(U)	78
Application(A)	12
<b>Total</b>	<b>100</b>

**Instruction for teacher/Paper setters:**

Teachers are requested to follow the above sample of specification while designing the test/paper



### **BIOLOGY GRADE-X**

<p style="text-align: center;"><b>Unit # 10</b> <b>GASEOUS EXCHANGE</b></p> <ul style="list-style-type: none"> <li>- Respiration, Gaseous Exchange and breathing</li> <li>- Gaseous exchange in plant</li> <li>- Gaseous exchange in Human</li> <li>- Mechanism of breathing</li> <li>- Gaseous exchange in Lungs</li> <li>- Respiratory Disorders</li> <li>- Pneumonia</li> <li>- Asthma</li> </ul>	<p style="text-align: center;"><b>Unit # 12</b> <b>COORDINATION AND CONTROL</b></p> <ul style="list-style-type: none"> <li>- Coordination in organisms (only 12.1)</li> <li>- Human Nervous System</li> <li>- Neuron, Nerves</li> <li>- Division of Nervous System (CNS) (Central Nervous System and peripheral Nervous system)</li> <li>- Receptors in the Human Body</li> <li>- Photoreceptor eye</li> <li>- Mechanism of vision</li> <li>- Disorder of the eye</li> <li>- Muslim Scientist in Ophthalmology (Ibn Al Haitham, Ali Bin Isa)</li> <li>- Chemical Coordination</li> <li>- Human endocrine gland (Pituitary gland, thyroid gland, pancreas, Adrenal gland, Gonads)</li> <li>- Disorders of Nervous System</li> <li>- Epilepsy (Fits or mirgi)</li> </ul>
<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>	
<p style="text-align: center;"><b>Unit # 11</b> <b>HOMEOSTASIS</b></p> <ul style="list-style-type: none"> <li>- Homeostasis the plants</li> <li>- Osmoregulation in plants</li> <li>- Mesophytes</li> <li>- Hydrophytes</li> <li>- Xerophytes</li> <li>- Halophytes</li> <li>- Excretion in plants</li> <li>- Excretion of Carbon dioxide and Oxygen</li> <li>- Excretion of extra water (transpiration and guttation)</li> <li>- Homeostasis in Humans</li> <li>- Lungs as homeostasis organs</li> <li>- Skin as homeostatic organ</li> <li>- Kidney as homeostatic organs</li> <li>- Urinary system of Humans</li> <li>- Structure of Human Kidney (Renal corpuscle, renal tubule)</li> <li>- Process of Urine formation (Pressure filtration, reabsorption, tubular secretion)</li> <li>- Osmoregulation and Kidney</li> <li>- Disorder of kidney</li> <li>- Kidney stone</li> <li>- Treatment of kidney stone</li> </ul>	<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>
<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>	<p style="text-align: center;"><b>Unit # 13</b> <b>SUPPORT AND MOVEMENT</b></p> <ul style="list-style-type: none"> <li>- Human Skeleton</li> <li>- Axial Skeleton</li> <li>- Appendicular Skeleton</li> <li>- Role of Skeletal System</li> <li>- Composition of skeleton (Bones, Cartilage)</li> <li>- Joints</li> <li>- Role of Tendons and Ligaments</li> <li>- Muscles and its types</li> </ul>
	<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>
	<p style="text-align: center;"><b>Unit # 14</b> <b>REPRODUCTION</b></p> <ul style="list-style-type: none"> <li>- Types Reproduction (Sexual and A Sexual reproduce)</li> <li>- A Sexual reproduction in Unicellular Organism and Plants</li> <li>- Binary fission (only in Amoeba)</li> <li>- Multiple fission</li> <li>- Budding</li> <li>- Sexual reproduction in plants</li> <li>- A Sexual reproduction in animals</li> </ul>



<b>BIOLOGY GRADE-X</b>	
<p style="text-align: center;"><b>Unit # 14</b> <b>REPRODUCTION</b></p> <ul style="list-style-type: none"> <li>- Fragmentation</li> <li>- Budding</li> <li>- Parthenogenesis</li> <li>- Sexual reproduction in animals</li> <li>- Formation of gametes – gametogenesis</li> <li>- Oogenesis</li> <li>- Fusion of gametes – Fertilization (External fertilization, Internal fertilization)</li> <li>- Need of population planning</li> <li>- Sexually transmitted disease</li> <li>- AIDS</li> </ul>	<p style="text-align: center;"><b>Unit # 16</b> <b>MAN AND HIS ENVIRONMENT</b></p> <ul style="list-style-type: none"> <li>- Interaction in the ecosystem (Competition, Predation)</li> <li>- Global and regional environmental problems (Deforestation and Global warming)</li> </ul>
<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>	<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>
<p style="text-align: center;"><b>Unit # 15</b> <b>INHERITANCE</b></p> <ul style="list-style-type: none"> <li>- Introduction to genetics</li> <li>- Chromosomes and genes</li> <li>- Chromatin material</li> <li>- Gene and allele</li> <li>- Role of gene in protein synthesis</li> <li>- Mendel’s work on inheritance</li> <li>- Mendel’s Law of inheritance</li> <li>- Law of segregation</li> <li>- Law of independent assortment</li> <li>- Variation and evaluation</li> <li>- Sources of variation (genetic recombination of chromosomes, crossing over, Mutation)</li> </ul>	<p style="text-align: center;"><b>Unit # 17</b> <b>BIOTECHNOLOGY</b></p> <ul style="list-style-type: none"> <li>- Importance of Biotechnology</li> <li>- Fermentation (Fermentation as a type of cellular respiration)</li> <li>- Alcoholic fermentation by yeast cells</li> <li>- Lactic acid fermentation by bacteria</li> <li>- Fermentation products</li> <li>- Genetic engineering and its uses</li> <li>- Objectives of genetic engineering</li> <li>- Basic techniques in genetic engineering</li> <li>- Major achievement of genetic engineering (Achievement in the field of agriculture, achievement in curing animal diseases)</li> </ul>
<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>	<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>
<p style="text-align: center;"><b>Unit # 16</b> <b>MAN AND HIS ENVIRONMENT</b></p> <ul style="list-style-type: none"> <li>- Levels of ecological organization. (Species, Population, Community, Ecosystem, Biosphere)</li> <li>- Components of an ecosystem</li> <li>- A biotic component of an ecosystem (Sunlight, Water, Temperature)</li> <li>- Biotic components of an ecosystem (Producers, Consumer, Decomposers)</li> <li><input type="checkbox"/> Food Chain and Food web</li> <li>- Biogeochemical Cycle</li> <li>- Carbon cycle</li> </ul>	<p style="text-align: center;"><b>Unit # 18</b> <b>PHARMACOLOGY</b></p> <ul style="list-style-type: none"> <li>- Medicinal drugs</li> <li>- Sources of medicinal drugs</li> <li>- Important uses of medicinal drugs (Pain killers, Vaccines)</li> <li>- Addictive drugs (Sedative, Narcotics)</li> <li>- Associated problems of drug addiction</li> <li>- Symptoms of drug addiction</li> <li>- Antibiotics</li> <li>- Categories of antibiotics (Sulfonamides, Tetracyclines, Cephalosporins)</li> <li>- Vaccine</li> </ul>
<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>	<p style="text-align: center;"><b>Exercise</b></p> <p>Only the questions given in exercise are related to these topics should be solved.</p>



<b>Biology Practical</b>	
1	Activity to compare the breathing rate at rest and after exercise
2	Examination of the structure of kidney (sheep kidney / model)
3	Observation and recording of the difference in quickness of response of the two types of coordination (by asking a student to say a few words in front of the class and observe the change in heartbeat).
4	Investigation of the nature of bone (by putting three pieces of rib bone of lamb in water, NaOH and dilute HCl)
5	Observation of binary fission of ameba using slides, photomicrographs or charts
6	Recording the heights of class fellows to predict which kind of variation is it and presentation of the data of class fellows' heights in graphical form (either histogram or bar chart)
7	Investigation of an ecosystem e.g. a balanced aquarium/pond
8	Investigation about the role of yeast and bacteria in the fermentation of flour and milk

**ASSESSMENT CRITERIA FOR FUTURE EXAMINATIONS:**

**Assessment weightage of Biology Grade –X According to (Curriculum, 2006)**

**SAMPLE: TOS (TABLE OF SPECIFICATION)**

**Table1. Cognitive abilities assessment weightage in Biology Grade-X Theory Paper**

Questions	Knowledge (K)			Understanding (U)			Application (A)		
	Questions	Marks	%	Questions	Marks	%	Questions	Marks	%
MCQs	2	2	17	9	9	75	1	1	8
Short Questions	1	4	10	9	36	80	1	4	10
Long Questions	1	3	11	3+3	9+12	74	1	4	15

**Table 2. Overall assessment Percentage of (Cognitive Levels)**

Cognitive Level	Percentage
Knowledge(K)	10
Understanding(U)	78
Application(A)	12
<b>Total</b>	<b>100</b>



**Table 3. Chapter wise assessment weightage in Biology Grade-X Theory Paper**

Unit No	Unit	Assessment weightage is given in Curriculum 2006	MCQs Section-A Marks: 12		Short Questions Section-B Option: 8 questions out of 11 Marks:32		Long Questions Section-C Option: 3 questions out of 4 Marks:21	
			Question	Marks	Questions	Marks	Questions	Marks
10	Gaseous Change	8	1	1	1	4	-	-
11	Homeostasis	12	1	1	1	4	1	3
12	Coordination and Control	17	2	2	2	8	1	3
13	Support and Movement	11	1	1	1	4	1	4
14	Reproduction	14	2	2	2	8	1	4
15	Inheritance	11	1	1	1	4	1	4
16	Man and His Environment	12	2	2	1	4	1	3
17	Biotechnology	7	1	1	1	4	1	3
18	Pharmacology	8	1	1	1	4	1	4
<b>Total</b>	-	<b>100</b>	<b>12</b>	<b>12</b>	<b>11</b>	<b>44</b>	<b>8</b>	<b>28</b>

